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## Key indicators

Single-crystal X-ray study
$T=291 \mathrm{~K}$
Mean $\sigma(1-\mathrm{Ag})=0.001 \AA$
Disorder in main residue
$R$ factor $=0.041$
$w R$ factor $=0.117$
Data-to-parameter ratio $=11.9$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## catena-Poly[4,4'-bipyridinium(+) [argentate(I)-di- $\mu$-iodo]]

In the crystal structure of the title compound, $\left\{\left(\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{~N}_{2}\right)\right.$ $\left.\left[\mathrm{AgI}_{2}\right]\right\}_{n}$, both the cation and anion adopt chain motifs. A hydrogen bond links the positively charged pyridinium end of one cation to the neutral pyridyl end of an adjacent cation; the anions are bridged by two I atoms into a zigzag chain. The Ag atom in the chain exists in a nearly regular tetrahedral environment, on a special position of site symmetry 2.

## Comment

Several di- $\mu_{2}$-iodo-argentate chain motifs with ammonium cations as counter-ions have been reported, for example, the tetramethylammonium (Peters et al., 1984), hexamethylethylenediammonium (Thackeray \& Coetzer, 1975), anilinium (Alcock et al., 2003) and ethane-1,2-dipyridinium (Li et al., 2004) salts. The present $4,4^{\prime}$-bipyridinium(+) analog is the crystalline product of the reaction between silver iodide and $4,4^{\prime}$-bipyridine in an attempted synthesis of the $4,4^{\prime}$-bipyridine adduct of silver iodide. The reaction of silver iodide and pyridine, on the other hand, yields the $1: 1$ silver iodide-pyridine adduct, which has a tetranuclear composition (Engelhardt et al., 1989).


The anionic chain of the title compound, (I), viz. catena-poly[argentate(I)-di- $\mu$-iodo] adopts an iodine-bridged chain motif (Fig. 1), the polyanion propagating as a zigzag chain along the $c$ axis of the monoclinic unit cell. The metal atom shows a nearly regular tetrahedral geometry, and the two independent Ag -I bonds are not statistically different from each other. The cation also adopts a linear chain motif, and one positively charged pyridinium end of the protonated cation interacts with the neutral pyridyl end of the adjacent cation $[\mathrm{N} \cdots \mathrm{N}=2.71$ (1) $\AA$ ]. The two rings are twisted by 27.4 (4) ${ }^{\circ}$ with respect to each other. The $\mathrm{N}-\mathrm{H} . \cdots \mathrm{N}$ hydrogenbonded cationic chain propagates in the [110] direction. Whereas the $4,4^{\prime}$-bipyridinium $(2+$ ) dication is relatively common, the $4,4^{\prime}$-bipyridinium $(+)$ monocation is rarely found in its metal salts, and only a few examples of these crystal structures have been reported (Al-Rasoul \& Weakley, 1982;

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Bukowska-Strzyzewska \& Tosik, 1978; Chippindale \& Turner, 1997; Hirose et al., 2002; Krishnamohan Sharma \& Rogers, 1999; Li et al., 2001, 2004; Liang et al., 2001; Rammo et al., 1994; Su et al., 1998; Weakley, 1984; Xu et al., 2002).

## Experimental

Silver iodide $(0.024 \mathrm{~g}, 0.1 \mathrm{mmol})$, potassium iodide $(0.100 \mathrm{~g}$, $0.6 \mathrm{mmol}), 4,4^{\prime}$-bipyridine ( $0.016 \mathrm{~g}, 0.1 \mathrm{mmol}$ ) and triphenylphosphite $(1 \mathrm{ml})$ were dissolved in a DMF/methanol mixture $(10 \mathrm{ml})$ to give a clear solution. Red crystals separated from the solution in about a week.

## Crystal data

$\left(\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{~N}_{2}\right)\left[\mathrm{AgI}_{2}\right]$
$M_{r}=518.86$
Monoclinic, $C 2 / c$
$a=13.6038$ (9) $\AA$
$b=13.9457$ (9) $\AA$
$c=7.1524$ (7) $\AA$
$\beta=103.031$ (1) ${ }^{\circ}$
$V=1322.0$ (2) $\AA^{3}$
$Z=4$

## Data collection

Bruker APEX-II area-detector
diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.128, T_{\text {max }}=0.373$
3879 measured reflections
$D_{x}=2.607 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 2830 reflections
$\theta=2.9-29.2^{\circ}$
$\mu=6.17 \mathrm{~mm}^{-1}$
$T=291$ (2) K
Block, red
$0.38 \times 0.27 \times 0.16 \mathrm{~mm}$

> 1463 independent reflections
> 1392 reflections with $I>2 \sigma(I)$
> $R_{\text {int }}=0.026$
> $\theta_{\max }=27.5^{\circ}$
> $h=-17 \rightarrow 16$
> $k=-16 \rightarrow 17$
> $l=-9 \rightarrow 9$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.041$
$w R\left(F^{2}\right)=0.117$
$S=1.10$
1463 reflections
123 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.066 P)^{2} \\
&+8.4861 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=1.71 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-1.09 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left(\AA^{\circ}{ }^{\circ}\right)$.

| $\mathrm{Ag} 1-\mathrm{I} 1$ | $2.8759(6)$ | $\mathrm{Ag} 1-\mathrm{I} 1^{\mathrm{i}}$ | $2.8882(6)$ |
| :--- | :---: | :--- | ---: |
| $\mathrm{I} 1-\mathrm{Ag} 1-\mathrm{I} 1^{\mathrm{i}}$ | $113.05(2)$ | $\mathrm{I} 1-\mathrm{Ag} 1-\mathrm{I} 1^{\mathrm{iii}}$ | 114.88 (3) |
| $\mathrm{I} 1-\mathrm{Ag} 1-\mathrm{I} 1^{i i}$ | 103.26 (1) |  |  |
| Symmetry codes: (i) $x,-y+1, z+\frac{1}{2} ;$ (ii) $-x+1,-y+1,-z ;$ (iii) $-x+1, y,-z+\frac{1}{2}$. |  |  |  |

The 4,4 $4^{\prime}$-bipyridinium( + ) cation is disordered over an inversion center, and it was refined as two pyridyl rings that are linked by a 1.50 (1) $\AA \mathrm{C}-\mathrm{C}$ bond. In both rings, the $\mathrm{N}-\mathrm{C}$ distances were restrained to 1.35 (1) $\AA$ and the $\mathrm{C}-\mathrm{C}$ distances to 1.39 (1) $\AA$. The rings were also restrained to be almost flat, the six atoms in each ring not being allowed to deviate from the plane by more than 0.01 A . Carbon-bound H atoms were placed at calculated positions $(\mathrm{C}-\mathrm{H}=$ $0.93 \AA$ ), and they were included in the refinement in the riding-model approximation, with $U_{\text {iso }}(\mathrm{H})$ values set at 1.2 times $U_{\text {eq }}(\mathrm{C})$. The positive charge was arbitrarily placed on atom N 1 (instead of atom N 2 ); the $\mathrm{N}-\mathrm{H}$ distance was fixed at $0.86 \AA$, and the displacement parameter was similarly tied to $U_{\text {eq }}(\mathrm{N})$. The final difference Fourier map had a large peak and a large hole, both at about $1 \AA$ from atom Ag 1 but was otherwise featureless.


Figure 1
ORTEP plot (Johnson, 1976) of a portion of the polymeric structure of (I). Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are drawn as spheres of arbitrary radii [symmetry codes: (i) $x$, $1-y, \frac{1}{2}+z$; (ii) $1-x, 1-y,-z$; (iii) $\left.1-x, y, \frac{1}{2}-z\right]$. The disorder is not shown.

Data collection: SMART (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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